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Key indicators

Single-crystal X-ray study T = 187 K Mean σ (C–C) = 0.003 Å R factor = 0.053 wR factor = 0.149 Data-to-parameter ratio = 12.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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6,7,8,9-Tetrahydro-5*H*-pyrido[2,3-*b*]azepine-5,8dione: a hydrogen-bonded dimer structure

In the title compound, $C_9H_8N_2O_2$, two crystallographically independent molecules form a dimer structure, in which two $N-H \cdots N$ hydrogen bonds generate an intermolecular $R_2^2(8)$ ring.

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Comment

The title compound, (I), is an intermediate in the synthesis of 1-azakepaullone (Kunick *et al.*, 2004), which is an inhibitor of glycogen synthase kinase-3 (Doble & Woodgett, 2003) and a potential antidiabetic drug (Wagmann & Nuss, 2001).



In the crystal structure of (I), two crystallographically independent molecules form a dimer through $N-H\cdots N$ hydrogen bonds (Fig. 1 and Table 1). These hydrogen bonds generate an $R_2^2(8)$ ring (Bernstein *et al.*, 1995). The dihedral angle between the two pyridine rings in the dimer is 7.6 (2)°. Atom C7 of the seven-membered ring has *endo* and *exo* conformations in the two independent molecules (Fig. 1).

The crystal structure consists of packed hydrogen-bonded dimers (Fig. 2). A π - π interaction is observed between the pyridine C10-C13/N2/C14 rings of neighbouring dimers; the centroid-to-centroid distance, $Cg \cdots Cg(-x, 2-y, -z)$, is 3.616 (2) Å and the interplanar spacing is 3.232 (2) Å.

Experimental

Compound (I) was prepared from 2-aminopyridine-3-carboxylic acid (purchased from Aldrich) according to the literature method of Kunick *et al.* (2003). Single crystals suitable for X-ray diffraction were obtained from an ethanol solution.

Crystal data	
$C_9H_8N_2O_2$	$V = 767.98 (18) \text{ Å}^3$
$M_r = 176.17$	Z = 4
Triclinic, P1	$D_x = 1.524 \text{ Mg m}^{-3}$
a = 8.2902 (11) Å	Mo $K\alpha$ radiation
b = 8.5284 (11) Å	$\mu = 0.11 \text{ mm}^{-1}$
c = 11.1261 (15) Å	T = 187 (2) K
$\alpha = 98.155 \ (2)^{\circ}$	Block, colourless
$\beta = 98.996 \ (2)^{\circ}$	$0.20 \times 0.13 \times 0.10$ mm
$\gamma = 91.754 \ (2)^{\circ}$	

Data collection

Bruker SMART CCD area-detector diffractometer φ and ω scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996) $T_{\min} = 0.978, T_{\max} = 0.989$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.053$ $wR(F^2) = 0.149$ S = 1.022959 reflections 235 parameters H-atom parameters constrained 4341 measured reflections 2959 independent reflections 2109 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.018$ $\theta_{\text{max}} = 26.0^{\circ}$

$$\begin{split} &w = 1/[\sigma^2(F_{\rm o}^2) + (0.0819P)^2 \\ &+ 0.0598P] \\ &where \ P = (F_{\rm o}^2 + 2F_{\rm c}^2)/3 \\ (\Delta/\sigma)_{\rm max} = 0.001 \\ \Delta\rho_{\rm max} = 0.23 \ {\rm e} \ {\rm \AA}^{-3} \\ \Delta\rho_{\rm min} = -0.28 \ {\rm e} \ {\rm \AA}^{-3} \end{split}$$

Table 1

Hydrogen-bond geometry (Å, $^{\circ}$).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$\begin{array}{c} N3 - H3A \cdots N2^{i} \\ N4 - H4A \cdots N1^{i} \end{array}$	0.88	2.21	3.051 (2)	161
	0.88	2.25	3.042 (2)	150

Symmetry code: (i) -x, -y + 2, -z + 1.

H atoms were found in a difference Fourier map and refined as riding, with C-H = 0.95–0.99 Å and N-H = 0.88 Å, and with U_{iso} (H) = $1.2U_{eq}$ (C,N).

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINT* (Bruker, 2003); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

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Figure 1

A view of the hydrogen-bonded dimer of (I), with the atom-labelling scheme and 30% probability displacement ellipsoids. Hydrogen bonds are shown as dashed lines. [Symmetry code (i) is as in Table 1].



Figure 2

A packing view of (I), along the a axis. Hydrogen bonds are shown as dashed lines. H atoms not involved in hydrogen bonding have been omitted for clarity.

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